

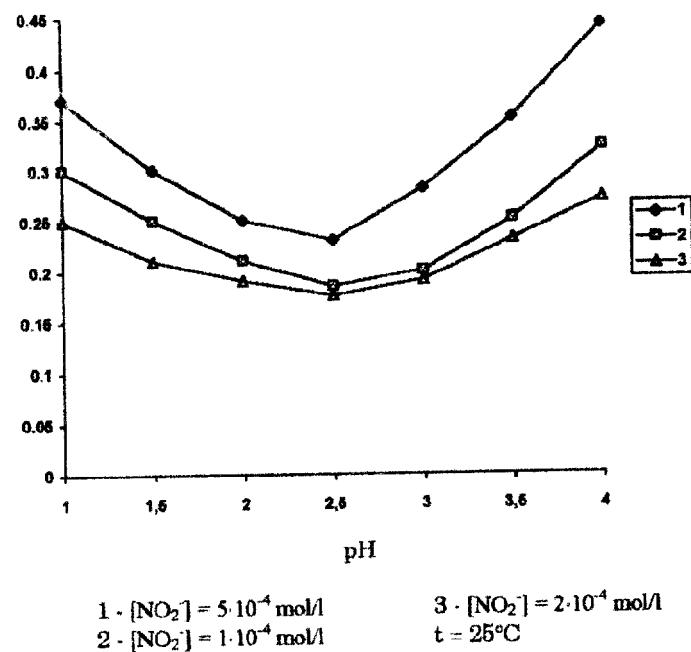
D(NO₂)230 nm

Fig. 1

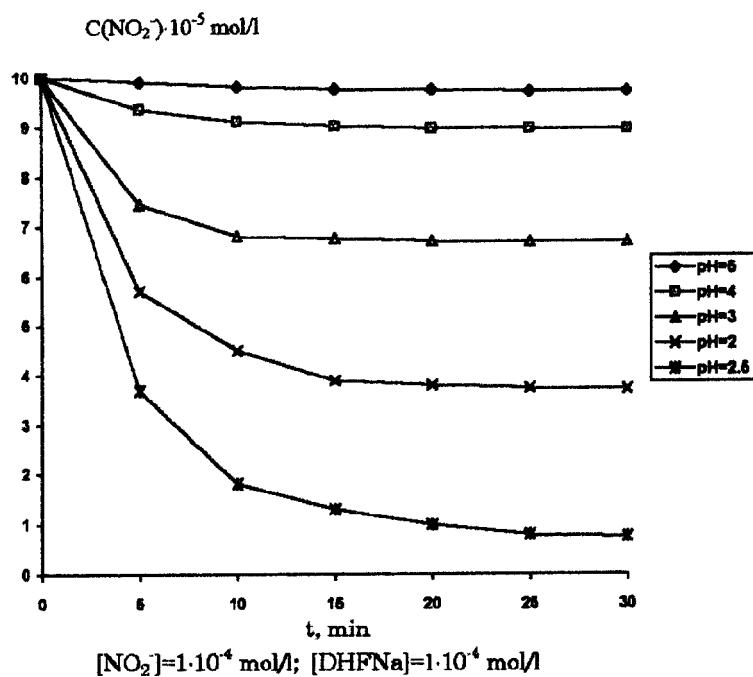


Fig. 2

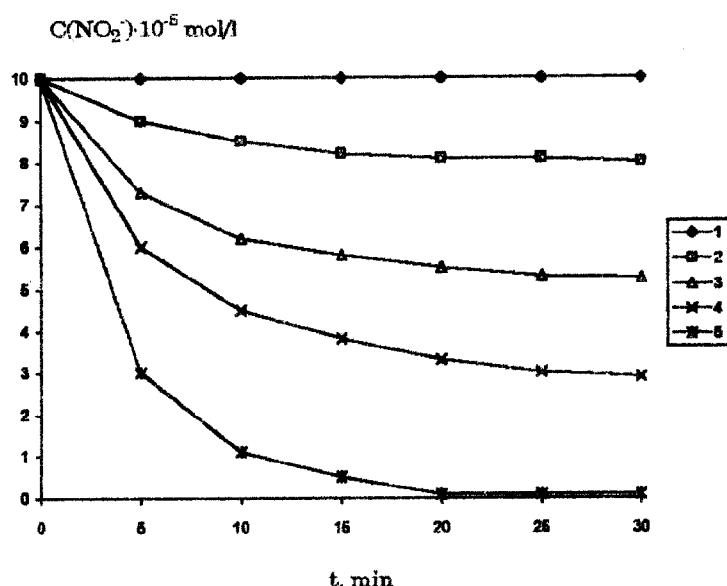


Fig. 3

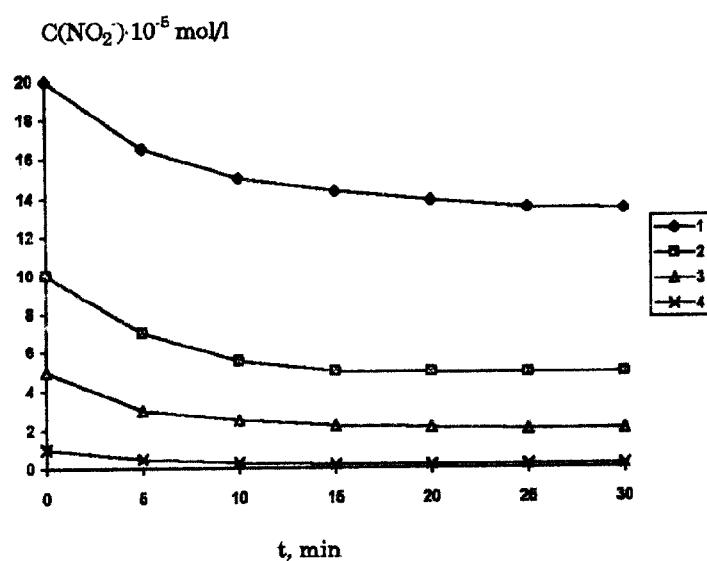


Fig. 4

$1 - [\text{NO}_2] = 2 \cdot 10^{-4} \text{ mol/l}$ $3 - [\text{NO}_2] = 5 \cdot 10^{-5} \text{ mol/l}$ $t = 25^\circ\text{C}$
 $2 - [\text{NO}_2] = 1 \cdot 10^{-4} \text{ mol/l}$ $4 - [\text{NO}_2] = 1 \cdot 10^{-5} \text{ mol/l}$ $\text{pH} = 2,5$
 $[\text{DHFNa}] = 5 \cdot 10^{-4} \text{ mol/l}$